TRIHYDROXY-C₁₈-ACIDS AND A LABDANE FROM RUDBECKIA FULGIDA

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Abstract—Extraction of Rudbeckia fulgida furnished 13α H-labd-8(17)-en-15-al-19-oic acid, two new C_{18} -acids tentatively formulated as 9 (S*),12 (S*)-13 (S*)-trihydroxyoctadeca-10 (E),15 (Z)-dienoic acid and 9 (S*),12 (S*),13 (S*)-trihydroxyoctadec-10 (E)-enoic acid, several known C_{14} -polyacetylenes and several flavone glycosides.

INTRODUCTION

Knowledge of the chemistry of Rudbeckia (Heliantheae) is relatively sparse [1-6]. As a result of our discovery of ambrosanolides with antitumour activity in R. mollis [6], we have begun a study of other Rudbeckia species. In the present article, we report the isolation from R. fulgida Ait.* of a new labdane 1a, the new trihydroxy acids 2a and 3a, the known acetylenes 5a-5c, 2,6-dimethoxybenzo-quinone and the flavone glycosides 6a-6e.

RESULTS AND DISCUSSION

Detailed analysis of the ¹H NMR spectra of the labdane, $C_{20}H_{32}O_3$ (high-resolution mass spectrometry), and its methyl ester led to the formulation of the parent acid as 1a without commitment as to stereochemistry. The diaxial relationship between the C-10 methyl and the C-4 carboxyl groups was evident from the upfield shift ($\Delta\delta 0.1$) of the C-10 methyl signal in going from 1a to 1b. To establish the stereochemistry at C-13 and the absolute stereochemistry of the molecule, 1a was converted to the dimethyl ester 1d. The ¹H NMR spectrum of the latter was identical to that reported for dimethyl oliverate [8]

Table 1. ¹H NMR spectral data of compounds 2b, 3b, 4b and 4d (270 MHz, CDCl₃)*

Н	2b	3b	4b	4d
 2†	2.30 t (7)	2.30 t	2.30 t	2.30 t
3†	1.59 m	1.59 m	1.60 m	1.62 m
4 7	1.30 br	1.28 br	1.30 br	1.29 br
8†	1.59 m	1.59 m	1.60 m	1.62 m
9	5.29 q (6)	5.21 q	5.24 q	5.24 q
10	5.69 dd (15, 6)	5.69 dd	5.75 dd	5.72 dd
11	5.58 dd (15, 6)	5.57 dd	5.64 dd	5.62 dd
12	5.37 t (6)	5.35 t	5.36 dd (6, 4)	5.34 dd
13	5.03 q (6)	5.03 q	5.04 ddd (8, 6, 4)	5.02 ddd
14†	2.30 (obs)	1.50 q (br) (7)	2.28 (obs)	1.49 q (br) (7)
15	5.24 (obs)	`	5.32 (obs)	
16	5.51 dtt	1.28 br	5.50 dtt	1.26 br
	(10.5, 7, 1.5)	}	1	
17†	2.04 quintet (7)	J	2.02 quintet	
18‡	0.96 t (7)	0.87 t (br)	0.97 t	0.88 t (br)
OMe‡	3.68	3.67	3.68	3.68
OAc‡	2.11, 2.07	2.08, 2.06	2.06, 2.06	2.06, 2.06
	2.07	2.05	2.04	2.06

^{*}Coupling constants (in parentheses) are not repeated if identical with those in preceding column.

^{*}Our collection of R. fulgida represented var. fulgida [7].

[†]Intensity two protons.

[‡]Intensity three protons.

but the rotation had the opposite sign. Hence 1a was formulated as 13αH-labd-8[17]-en-15-al-19-oic acid.

The two acids 2a and 3a could be separated with some difficulty only in the form of their derivatives, 2b and 3b. Sequential spin decoupling of the ¹H NMR spectra of these compounds (Table 1) and mass spectrometry (Schemes 1 and 2) permitted their formulation as a methyl 9,12,13-triacetoxyoctadeca-10(E),15(Z)-dienoate and a methyl 9,12,13-triacetoxyoctadeca-10(E)-enoate, respectively. We have named the parent compound (2a) fulgidic acid.

The structure and absolute stereochemistry, 4a (9S,12R,13S), have been deduced recently [9] for malyngic acid from the blue-green alga Lyngia majuscula.* Direct

comparison of the ¹H NMR spectra of 2b and 3b with the spectra (see Table 1) of authentic methyl malyngate triacetate (4b) and methyl dihydromalyngate triacetate

OAC OAC
$$m/z$$
 468 -31 437(0 3)
$$-59 -60 -42 -60
226(100) 208(17) -42 -60
306(80) 288(11)$$

Scheme 1.

^{*}A malyngic acid sample kindly supplied by Professor R. B. Moore actually was a mixture of malyngic acid and its 15,16-dihydro derivative 4c. Separation of 4b and 4d after methylation and acetylation was achieved as detailed for 2c and 3c in the Experimental.

Scheme 2.

(4d) clearly demonstrates that the two sets of compounds differ in relative stereochemistry. From the difference in coupling constants involving H-12 and H-13, we infer that fulgidic acid and its 15,16-dihydroderivative are 12,13threo-diols rather than erythro-diols like 4a and 4c. Moreover, the significant differences in the chemical shifts of H-10, H-11 and H-15 between 2b and 4b and H-10 and H-11 between 3b and 4d suggest that the spatial relationship between H-9 and H-12 is altered and that fulgidic acid and its dihydro derivative therefore possess the relative stereochemistry 9S*,12S*,13S*. Unfortunately, authentic samples of 9(S),12(R),13(S)9(S),12(S),13(S)-trihydroxystearic acids [9] with one of which we had hoped to correlate 2a and 3a were no longer available, so that this proposal must remain tentative.

The results of the present study reinforce the impression that the chemistry of this genus is not very uniform, but further work is clearly needed.

EXPERIMENTAL

Above-ground parts (4.9 kg) of Rudbeckia fulgida Ait. var. fulgida, collected by Dr. John L. Nelson on 25 September 1982 along U.S. 98, approximately 2 miles east of the Aucilla River, Taylor Co., Florida (Nelson and Wnek 2281 on deposit at the herbarium of the University of South Carolina), were extracted with CHCl₃ and worked up in the usual fashion [10]. The crude gum (29 g) was absorbed on 40 g silicic acid (Mallinckrodt 100 mesh) and chromatographed over 600 g of the same absorbent packed in hexane, 500 ml fractions being collected as follows: 1-2 (hexane), 3-8 (hexane-EtOAc, 19:1), 9-14 (hexane-EtOAc, 9:1), 15-18 (hexane-EtOAc, 4:1), 19-22 (hexane-EtOAc, 3:2), 23-26 (hexane-EtOAc, 7:1), 27-30 (hexane-EtOAc, 2:3), 31-34 (hexane-EtOAc, 1:4), 35 and 36 (EtOAc), 37 and 38 (EtOAc-MeOH, 49:1), 39-42 (EtOAc-MeOH, 19:1) and 43-46 (EtOAc-MeOH, 9:1).

Purification of fraction 5 by TLC (C_6H_6 -EtOAc, 39:1) afforded 62 mg 5a, MS m/z (rel. int.): 240 [M]⁺ (99), 198 (4.5), 197 (24), 181 (22), 180 (66), 179 (71), 178 (61.5) and 165 (100); ¹³C NMR spectrum (67.89 MHz, CDCl₃): 4.66 (q, C-1), 78.57, 76.62, 75.21, 68.04, 65.00, 59.26 (all s, C-2 to C-7), 107.46 (d, C-8), 146.27 (d, C-9), 130.19 (d, C-10), 138.63 (d, C-11), 29.24 (t, C-12), 27.84 (t, C-13), 63.67 (t, C-14), 171.03 (s) and 20.91 (q, OAc). All multiplets were assigned by single frequency decoupling.

Fraction 7 after esterification with CH_2N_2 and purification by TLC (C_6H_6 -EtOAc, 39:1, multiple development) yielded methyl stearate (52 mg) and methyl oleate (25 mg). Fractions 11 and 12 contained mainly one substance; purification by TLC (C_6H_6 -EtOAc, 19:1) afforded slightly impure 1a (54 mg); MS m/z (rel. int.): 320 [M]⁺ (3), 274 (8), ¹H NMR (270 MHz,

CDCl₃): δ 9.76 (t, J = 3 Hz, H-15), 4.84 (br) and 4.49 (br, H-17a, b), 1.25 (H-18), 0.9 (d, J = 6.5 Hz, H-16) and 0.60 (H-20). [Calc. for C₂₀H₃₂O₃: MW, 320.2351. Found: MW(MS), 320.2351]. Methylation gave spectroscopically pure **1b** as a gum, ¹H NMR (270 MHz, CDCl₃): δ 9.76 (t, J = 3 Hz, H-15), 4.87 (br) and 4.49 (br, H-17a, b), 3.63 (OMe), 1.18 (H-18), 0.97 (d, J = 6.5 Hz, H-16) and 0.50 (H-20). On standing, **1b** underwent air oxidation to **1c**, which was esterified to **1d** (gum), [α]_D + 44° (CHCl₃); IR ν CHCl₃ cm⁻¹: 1715 br; ¹H NMR: δ 4.83 (br) and 4 48 (br, H-17a, b), 3.66, 3.61 (OMe), 1.18 (H-18), 0.94 (d, J = 6 Hz, H-16) and 0.50 (H-20); MS m/z (rel. int.): 364 [M]⁺, 349 (1), 305 (15.4), 304 (27.7), 289 (4.3). [Calc. for C₂₂H₃₆O₄: MW, 364.2614. Found: MW(MS), 364.2643.]

Purification of fraction 13 by TLC (C_6H_6 -EtOAc, 9:1) gave 20 mg **5b**; MS m/z (rel. int.): 298 [M] $^+$ (6.6), 239 (14.5), 238 (65), 197 (14), 196 (70.5), 195 (87), 179 (55.5), 178 (100), 177 (41.2), 176 (18.7) and 165 (48). TLC of fraction 17 similarly furnished 32 mg **5c**. Fraction 25 on standing in hexane-EtOAc gave 45 mg 2,5-dimethoxybenzoquinone.

Purification of fraction 33 by TLC (CHCl₃-MeOH-EtOAc, 8:1:1) gave a fraction which appeared to be homogeneous but whose NMR spectrum showed it to be a mixture of closely related compounds. Esterification (CH₂N₂) and TLC effected no separation Acetylation in the usual way and separation of the triacetates by TLC (7% AgNO₃ on silica gel, C_6H_6 -EtOAc, 9:1, 2 developments) gave 15 mg 2b and 9 mg 3b as gums 1 H NMR: see Table 1; MS: see Schemes 1 and 2.

Fractions 42 and 43 on standing in EtOAc deposited 0.160 g quercitrin (6a). Fraction 44 on purification by TLC (CHCl₃-MeOH-EtOAc, 8:1:1) gave a mixture of flavone glycosides. Prep. TLC of the derived acetates (CHCl₃-MeOH-EtOAc, 18:1:1) gave two bands. The upper band, on repurification by the above solvent system (2 developments) gave in the upper band a mixture of quercitrin heptaacetate (acetate of 6a) and astragalin heptaacetate (acetate of 6d), and in the lower band, a mixture of 6c heptaacetate and populnin heptaacetate (acetate of 6e). Repurification of the lower band from the initial TLC purification gave in the upper band a complex mixture of the above compounds; the lower band yielded isoquercetrin octaacetate (acetate of 6b).

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REFERENCES

- Carman, N. J., Watson, T., Bierner, M. W., Averett, J., Sanderson, S., Seaman, F. C. and Mabry, T. J. (1982) Phytochemistry 11, 3271.
- Thompson, W. R., Meinwald, J., Aneshansley, D. and Eisner, T. (1972) Science 177, 528.
- Iyengar, M. A., Singri, B. P., Wagner, H., Seligman, O. and Herz, W. (1976) *Indian J. Chem. Sect. B* 14, 906
- Bohlmann, F., Jakupovic, J. and Zdero, C. (1978) Phytochemistry 17, 2034.
- Jauhari, P. K., Sharma, S. C., Tandon, J. S. and Dhar, M. M. (1979) Phytochemistry 18, 359.
- Herz, W., Kumar, N. and Blount, J. F. (1981) J. Org. Chem. 46, 1356.
- 7. Perdue, R. E., Jr. (1957) Rhodora 59, 293.
- Haeuser, J., Hall, S. F., Oehlschlager, A. C. and Ourisson, G. (1970) Tetrahedron 26, 3461.
- Abbot, G. G. and Gunstone, F. D. (1971) Chem. Phys. Lipids 7, 279.
- 10. Herz, W. and Högenauer, G. (1962) J. Org. Chem. 27, 905.